Brief Hydrogen Tutorial: Why Hydrogen, Why Now?

California Hydrogen Highway Network Media Hydrogen Workshop

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September 1, 2005



Outline

- Introduction to hydrogen technologies
- Hydrogen properties
- Current hydrogen uses
- Energy context and interest
- Issues and potential for expanded hydrogen use
- Summary



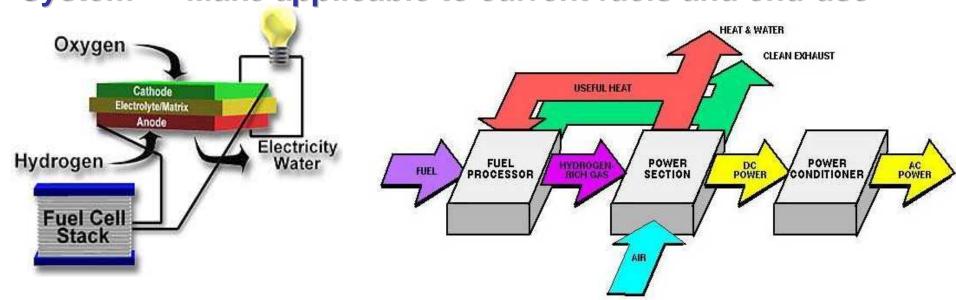
FUEL CELL CONCEPT

Electrode/Electrolyte Assemblies (similar to battery)
Continuous electricity production

Fuel and oxidant provided to separate chambers

"Stack" - Increase Voltage/Current to Useful Levels

"System" - Make applicable to current fuels and end-use



Primary Advantages – low, to zero emissions; high efficiency Primary Disadvantage – high cost (compared to current engines)

Fuel Cell Applications

STATIONARY /
DISTRIBUTED
POWER



Residential/Commercial/Industrial Appl.

Backup Power

Remote Power

Premium Power

TRANSPORTATION: HEAVY / LIGHT DUTY VEHICLES



Buses / Trucks

Passenger Vehicles

Locomotives / Shipping

PORTABLE POWER
BATTERY
REPLACEMENT



Consumer Electronics

Laptop Computers

Cellular Phones

Camcorders, etc.

Small Motors

Boating

Lawn Mower

Tools, etc.

FC DISTRIBUTED GENERATION

Stage: Initial Commercialization

- **High-cost Local Power**
 - e.g., 25-30 Million Homes (> \$0.10 / kWhr)
 - @ \$0.07-0.08/kWhr, \$500 Annual Savings



Most have natural gas connection

Possible Developing World Market

- **No Electricity 2 Billion People**
 - **DG** Paradigm may offer cost savings (cellular phone analogy)





Idatech







FuelCell Energy



UTC Fuel Cells



Siemens Westinghouse



Hydrogenics





Ballard-Ebara



FC TRANSPORTATION

Buses

Passenger Vehicles
/ Light Trucks

Electric Bikes

Golf Carts

Trucks

Shipping/Submarines

Locomotives

Snowmobiles

Aircraft

Other



TOYOTA FCHV



NEBUS



SCHATZ ENERGY RESEARCH CENTER



BOEING



MANHATTAN SCIENTIFICS



BWXT – McDermott U.S. Navy



BURLINGTON NORTHERN



FORD FOCUS







FC TRANSPORTATION



Nissan

Jeep Commander



DaimlerChrysler Necar4



Toyota, RAV4 (PEMFC)







FC TRANSPORTATION



Ford Model U



Daihatsu MOVE



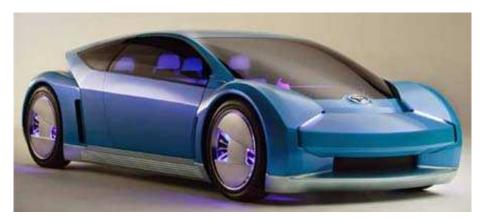
DaimlerChrysler – Necar5



GM-Opel FedEx FCV



Honda FCX



Toyota Fine S



FC TRANSPORTATION



GM Opel Hydrogen 1



Honda FCX V3



DaimlerChrysler & Ford



Mazda Demio



Nissan FCV



Toyota FCHV



Ford P2000 H2



BMW 700 Series



GM Precept



FC BUSES



DaimlerChrysler NeBus



Ballard - Vancouver



UTC Fuel Cells – Georgetown Bus



Mercedes-Benz Citaro 30

FC BUSES



Citaro FC Bus



Ballard – Chicago Bus Fleet



DaimlerChrysler NeBus



UTC Fuel Cells – Recent Georgetown Bus



OTHER FUEL CELL VEHICLES



Coval H2



Energy Partners



Energy Partners



Manhattan Scientifics



H Power Corp.



Outline

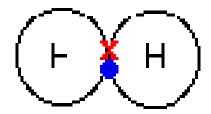
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13/40

HYDROGEN

- The lightest element
- Usually present as an H₂ molecule: "Diatomic Molecule"
- Two protons and two electrons
 - The two atoms (protons) share the pair of valence electrons leading to a STABLE molecule:

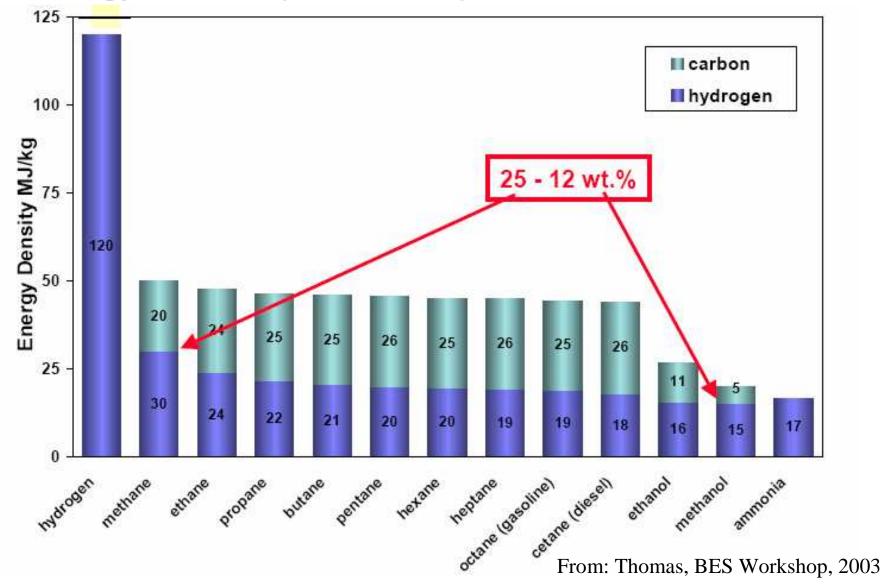


- Colorless, Odorless, Tasteless
- Flammable
 - Higher heating value (HHV) = 60,958 BTU/lb (141,670 kJ/kg)



HYDROGEN

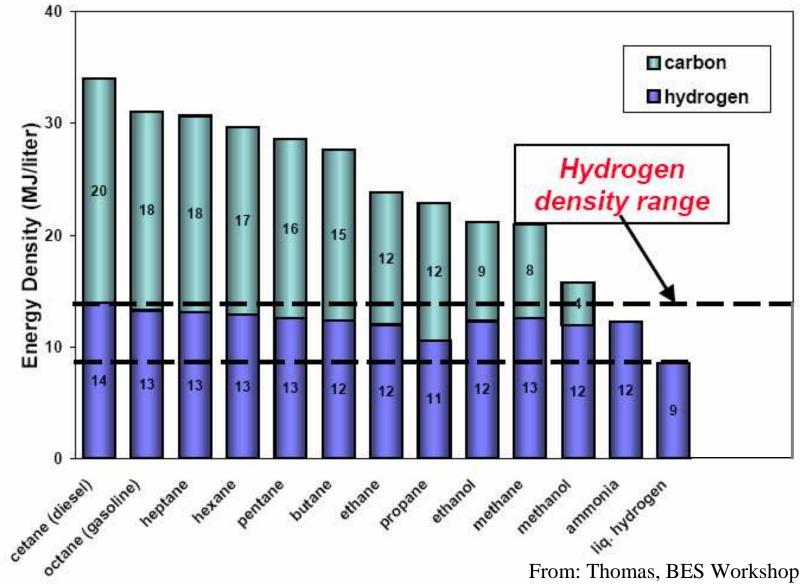
Energy Content (mass basis)





HYDROGEN

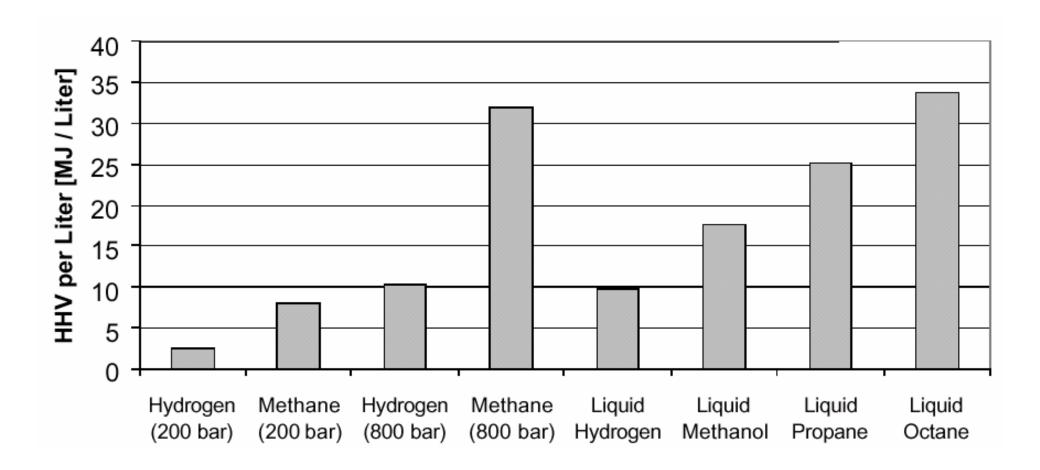
Energy Content (volume basis – liquid state)





HYDROGEN

Energy Content (volumetric basis)



From: Eliasson and Bossel, 2002



<u>HYDROGEN</u>

- Flammability (or inflammability)
 - Measured in terms of lowest and highest concentrations of the species that when premixed with air can sustain a flame in a tube or spherical container 2 inches in diameter
 - Indication of how easy it is to burn in various mixtures with air

Flammability Limits

% (by volume)

	Lower Limit	Upper Limit
Hydrogen:	4.65	93.9
Methane:	5.00	15.0
Propane:	2.12	9.35
Decane:	0.77	5.35
Benzene:	1.40	7.10



Other Hydrogen Properties

- High diffusivity (e.g., 3.8 times faster than natural gas)
 - when released, H₂ dilutes quickly into a non-flammable concentration
- Low density / high buoyancy
 - H₂ rises 2 times faster than helium and 6 times faster than natural gas
- H₂ flames have low radiant heat
 - risk of secondary fires is lower
- Ignition energy (in region of interest) is similar to other fuels
- An explosion cannot occur in a H₂ tank or any contained location that contains only hydrogen
 - an oxidizer, such as oxygen, must be present (not likely to explode)
- Asphyxiation can occur with H₂, but, its buoyancy and diffusivity make it difficult to confine (low asphyxiation risk)
- H₂ is non-toxic and non-poisonous
- Very significant advances in H₂ Codes and Standards that are in place for guiding safe handling, building and installation practices (especially handling as an energy carrier / fuel)

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Current Hydrogen Use

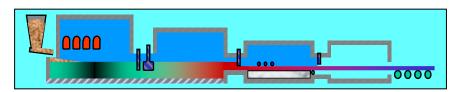
Current Hydrogen Use

- Hydrogen is widely used today (> 90 billion ft³/year)!
- Used mostly as a chemical, rather than a fuel, in a variety of commercial applications
 - Petroleum refining processes
 - Chemical process raw material (e.g., plastics, food-grade oil, ammonia)
 - Used as a reducing gas in metals processing (e.g., steel)
 - Electronics industry (e.g., silicon wafers and computer chips)
 - Rocket engine fuel (e.g., Space Shuttle)















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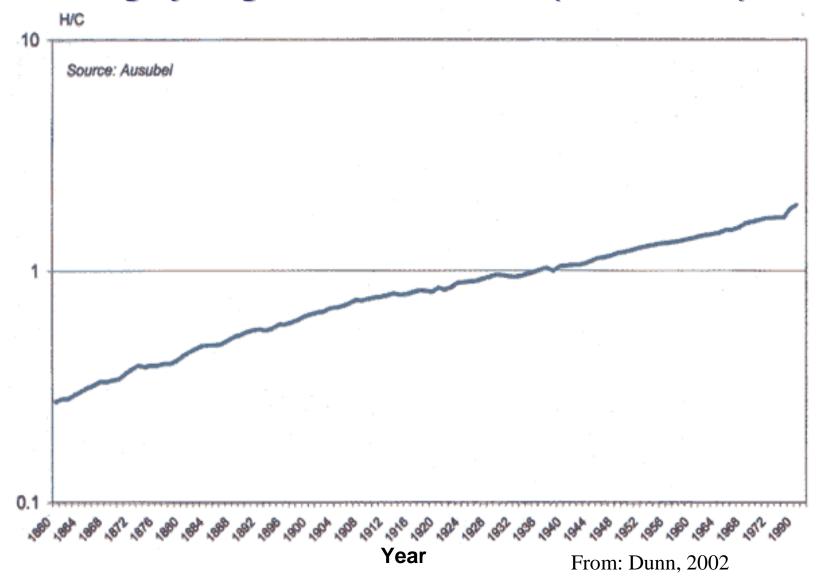
<u>History</u>

- Jules Verne 1874 novel "The Mysterious Island"
 - Idea of an energy system based on water and hydrogen
- Phrase 'hydrogen economy' coined in 1970 by GM engineers
- Four decades of relative neglect
- Recent interest and upheaval due to several factors
 - Security
 - Environment
 - Technology advances
 - Policy, etc.
- Is the energy world returning to chaos of early 20th century?
 - New transportation fuels and technologies
- Hydrogen related innovations are now important elements of energy industry strategy
 - Annual turnover of \$2 trillion



Fuel Hydrogen Content

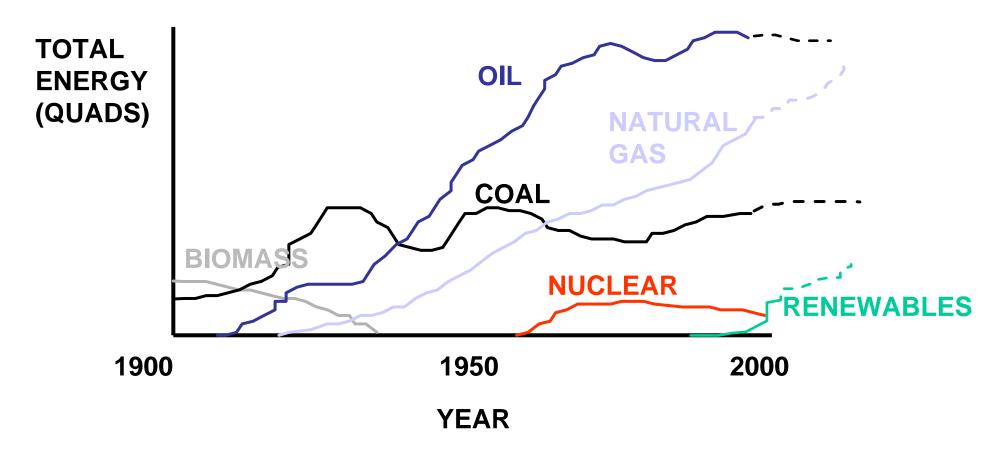
Increasing hydrogen content in fuels (1860 – 2000)





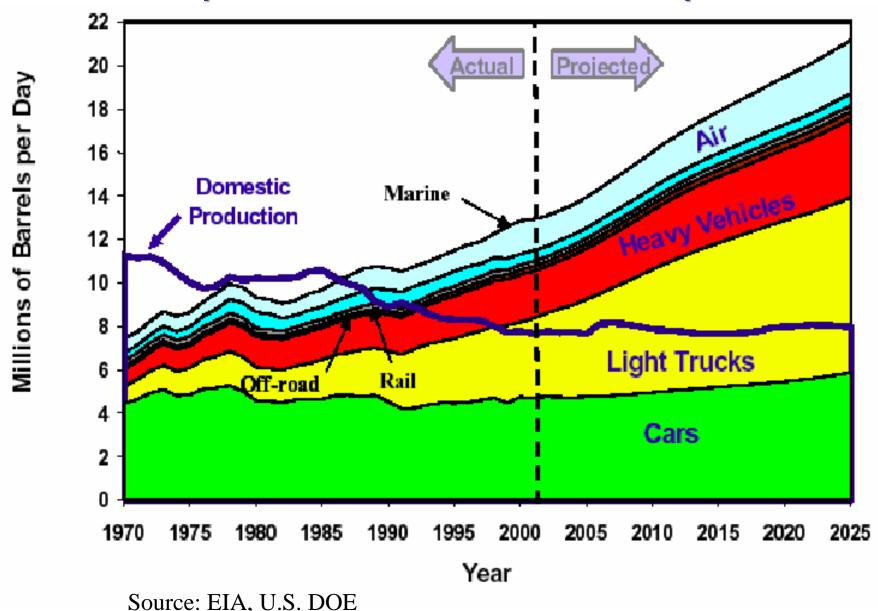
Energy Use

Worldwide Historical Fuel Use to Meet Energy Demands





Domestic oil production and vehicle consumption





Why use Hydrogen?

ENVIRONMENT:

- Conversion to thermal or electrical energy produces only water as a by-product (H₂ + ½ O₂ → H₂O)
- Compared to HC combustion: NO_x, SO_x, CO, HC, CO₂, lower
- Conversion devices may be more energy efficient (use less fuel, less emissions per unit of useful work): e.g., fuel cell

ENERGY SECURITY, INDEPENDENCE:

- Hydrogen can be made from domestic sources of primary energy (energy security, independence):
 - Natural gas

- Oil
- Electricity (renewable or grid)
- Coal

ECONOMICS:

- Cheap sources of fuel may be running out (~20-50 years)
 - Oil

Natural gas



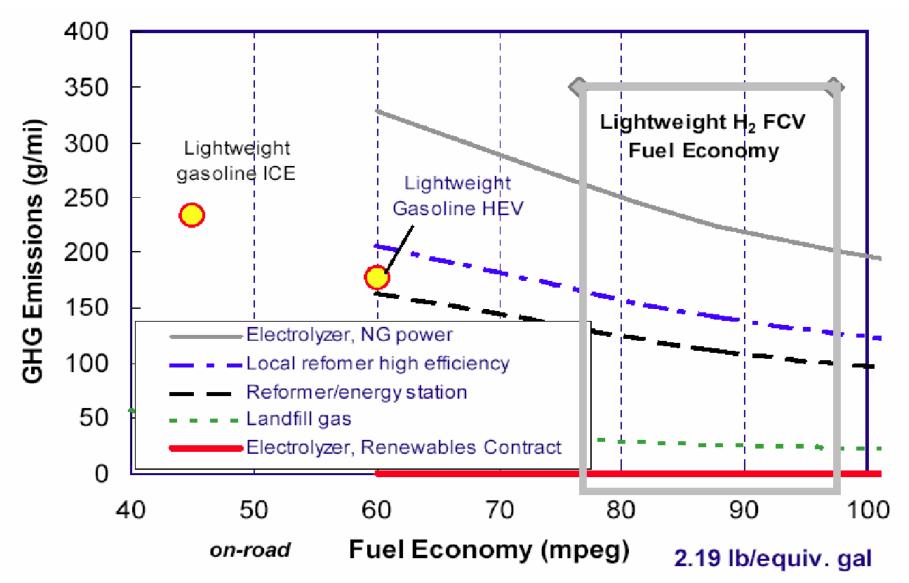
Why use Hydrogen?

 Limits to Environmental Sensitivity of Current Technologies (e.g., Automobile Engine)

HC EMISSIONS 5500 CO EMISSIONS NO_x EMISSIONS 22 5000 No Control 1300 No Control 20 Program No Control 1200 4500 Program EMISSIONS (Thousand Tons/Day) Program EMISSIONS (Tons/Day) (Tons/Day) 100 18 4000 000 16 3500 900 14 800 3000 **EMISSIONS** 12 700 2500 10 600 2000 500 400 1500 300 1000 200 500 100 1980 1990 1950 1960 1940 1950 1960 1970 1980 1990 1940 1950 1960 1970 1980 1990 YEAR END YEAR END YEAR END



GHGs for Hydrogen FCVs vs. ICE Vehicles



Source: Bevilacqua-Knight, 2001



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Issues for Increased H₂ Use - Safety

HYDROGEN SAFETY



Fire



Excessive Tank Pressure (Blocking all safety valves)



Mechanical Damage

Extensive testing & safety standards applied to all hydrogen apparatus



Hydrogen Leak

Gasoline Leak

Courtesy: BMW Group, 2000; Garrity, Murdoch Univ., 2002

Issues for Increased H₂ Use - Safety

HYDROGEN SAFETY

H₂ characteristics

- Broadest flammability limits
- Highest diffusivity
- Lowest density

Need to handle carefully, but, can be safer than gasoline!

 In the event of an accident – creation of a flammable mixture is less likely with hydrogen than with gasoline

H₂ is challenging to handle and requires consistent application of best practices

But – there is no significant technological hurdle Rather there is a need for education at several levels

- Engineers, designers, builders (codes and standards)
- End-Users
- General Public



Issues for Increased H₂ Use - Storage

High pressure hydrogen tank

Hydrogen-absorbing alloys/hydrides tank

Liquid hydrogen tank





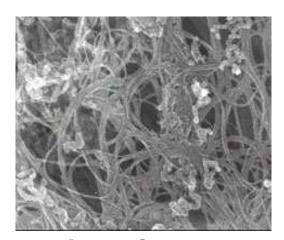


Issue: Volume

Weight

Boil-off gas

Carbon nanotubes



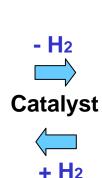
Issue: Actual storage

Chemical hydrides

NaBH4







Decalin($C_{10}H_{18}$) Naphthalene($C_{10}H_{8}$)



Handling / Recycling

Courtesy: Kawai, Toyota, 2002

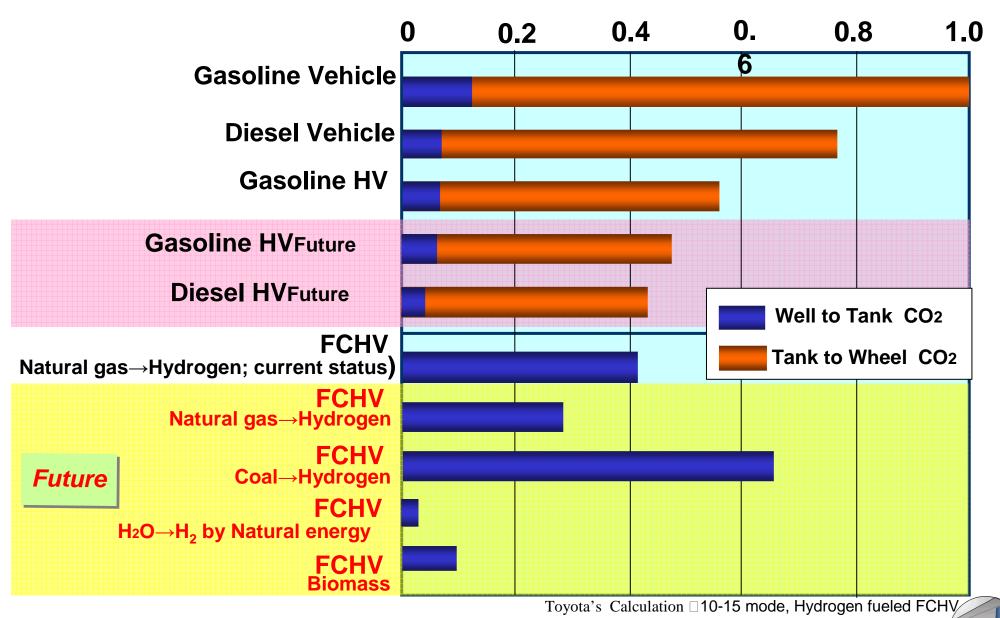
"Well- to- Wheel" Fuel Cycle Emissions

EXAMPLE:

- Extraction of Original Resource (e.g., in Saudi Arabia)
- Processing of Bulk Fuel (e.g., in Saudi Arabia)
- Storage of Bulk Fuel (e.g., in Saudi Arabia)
- Transportation of Bulk Fuel (e.g., in oil tanker)
- Bulk Storage (e.g., in California refinery)
- Production of Refined Product (e.g., in California refinery)
- Transportation and Distribution (e.g., to filling stations)
- Refined Product Storage (e.g., in filling station)
- Vehicle Storage (e.g., evaporative emissions)
- Vehicle Use (e.g., combustion emissions)

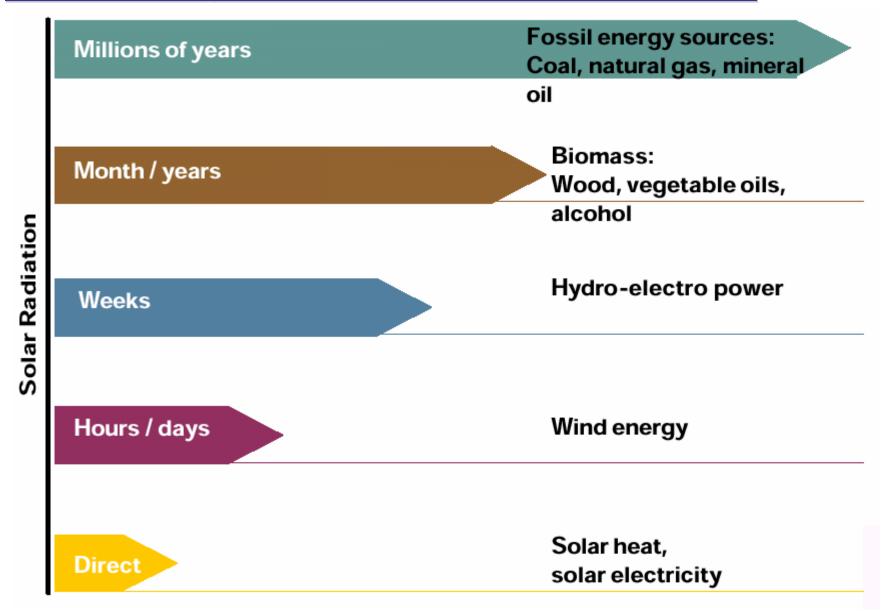
VERY IMPORTANT TO INCLUDE "WELL-TO-WHEEL" OR "LIFE-CYCLE" IMPACTS IN ANALYSES OF ALTERNATIVE FUELS





Courtesy: Reingit, Toyota, May, 2003

Primary Energy: All We Use Comes from the Sun



NFCRC

Courtesy: BMW Group, 2000

Societal "choices" that impact Life Cycle

- Primary energy used for hydrogen production
- Hydrogen transportation technology
 - Pipeline, wires, ...
 - Ships, trains, trucks, ...
- Location of hydrogen infrastructure
- Storage technology
 - Pressurized
 - Liquid
 - other
- End-use technology
 - Fuel Cells
 - Internal combustion engine
- Sustainability requires that we use resources (produce products) at the same *rate* at which they are naturally replenished (consumed)



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Summary – Why Hydrogen, Why Now?

- Hydrogen is one of only a few options (others: biofuels, battery EVs/plug-in HEVs) that can simultaneously address
 - Air pollution
 - Greenhouse gas (GHG) emissions
 - Energy and national security concerns
 - Energy sustainability (economic and environmental)
 - climate change
 - "peak oil" production
 - high prices
 - geopolitical concerns
- Great technological progress has been made (last 15 years)
 - Hydrogen technology
 - Fuel cell technology
 - Electric drive trains and Hybrid vehicles
- Challenges remain
 - Life Cycle, Cost (FCV, H₂ infrastructure), H₂ Storage



Thank You for Your Attention!

